

# Industrial Hydrogen Production & Technology

Linde Engineering

The Linde logo is a stylized, white, cursive script of the word "Linde" set against a dark blue background. The letters are elegant and flowing, with a slight shadow effect that makes them stand out.

*Linde*

Dr. Klemens Wawrzinek, Claude Keller  
HDV, November 21, 2007,  
Karlsruhe, FuncHy-Workshop

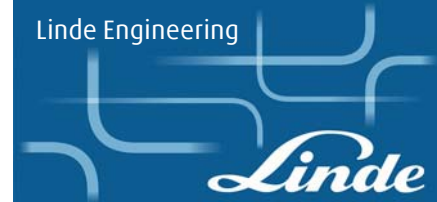
## Overview

- Linde Engineering's Key Plant Types
- Hydrogen Market
- Feedstocks
- Technology: Syngas Generation, Product Recovery
- Summary

# Linde Engineering Key Plant Types



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## Olefin Plants



Products:

- Ethylene
- Propylene
- Butadiene
- Aromatics
- Polymers

## Natural Gas Plants



— Products:

- LNG
- NGL
- LPG
- Helium

## Hydrogen and Synthesis Gas Plants



— Products:

- H<sub>2</sub>/CO/Syngas
- Ammonia
- Gas removal
- Gas purification

## Air Separation Plants



— Products:

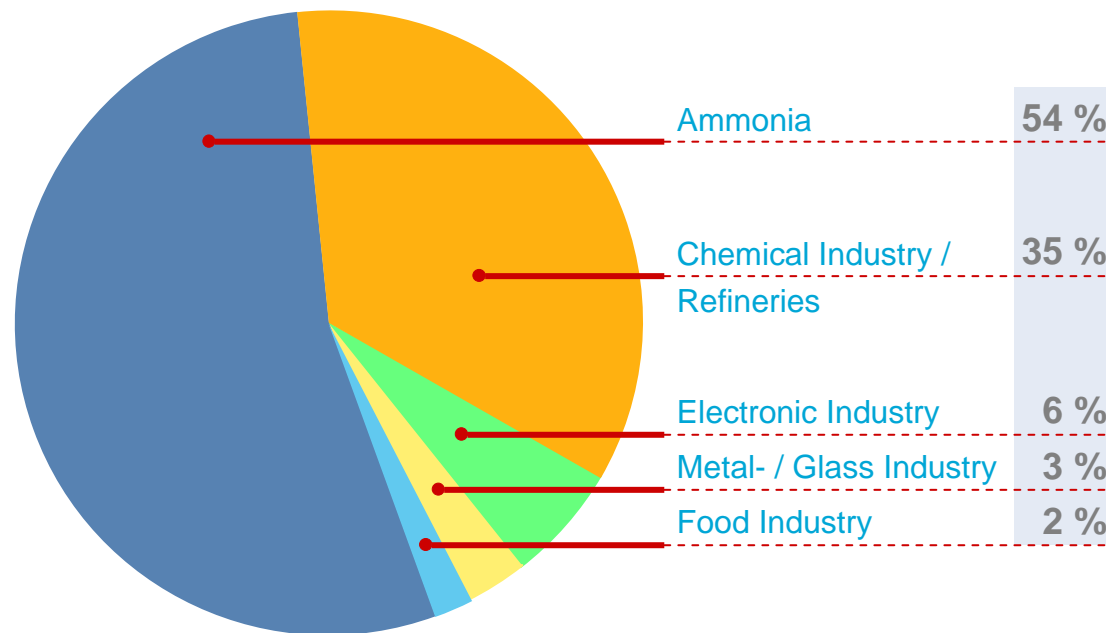
- Oxygen
- Nitrogen
- Rare gases

Installed capacity worldwide:

600 Billion Nm<sup>3</sup>/year

Hydrogen Consumers:

Trends shaping future Hydrogen demand:



- Increase of World Oil Consumption
- Decline of Overall Crude Oil Quality
- More Stringent Environmental Standards
- New Applications (Automotive fuel, Fuel cell)

## Light Hydrocarbons

- Refinery Gases
- LPG (Propane, Butane)
- Natural Gas (48 %)
- Naphtha

## Process

- Steam Reforming
- Partial Oxidation

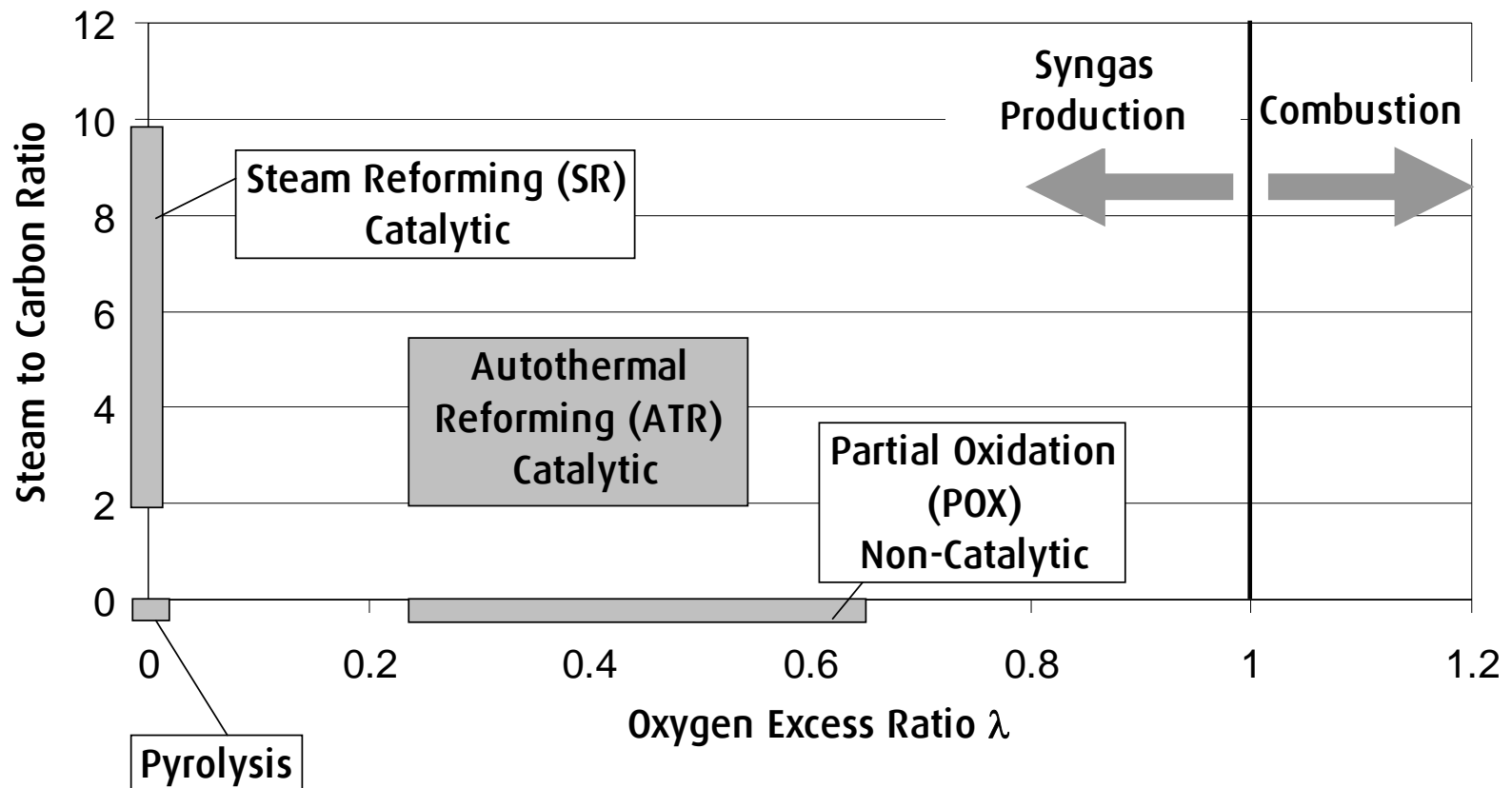
## Heavy Hydrocarbons

- Fuel Oil (30 %)
- Vacuum Tar
- Asphalt
- Petroleum Coke
- Coal (18 %)

## Process

- Partial Oxidation

# Synthesis Gas Generation Principles



## Non Oxygen Consuming:

- Steam Methane Reforming (SMR)  
 $\text{CH}_4 + \text{H}_2\text{O} \rightarrow \text{CO} + 3 \text{H}_2$       endothermal
- Carbon Monoxide Conversion (CO-Shift)  
 $\text{CO} + \text{H}_2\text{O} \rightarrow \text{CO}_2 + \text{H}_2$       exothermal

} Steam Reforming

## Oxygen Consuming

- Hydrocarbon Conversion  
 $\text{C}_n\text{H}_m + n/2 \text{O}_2 \rightarrow n\text{CO} + m/2 \text{H}_2$       exothermal
- H<sub>2</sub> Oxidation  
 $2 \text{H}_2 + \text{O}_2 \rightarrow 2 \text{H}_2\text{O}$       exothermal
- Carbon Monoxide Oxidation  
 $2 \text{CO} + \text{O}_2 \rightarrow 2 \text{CO}_2$       exothermal

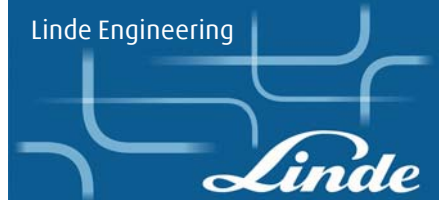
} Partial Oxidation, Autothermal Reforming

- **Synthesis Gas contains H<sub>2</sub>, CO, H<sub>2</sub>O, CO<sub>2</sub>, unreacted Hydrocarbons, Impurities**
- **Requested Products are H<sub>2</sub>, CO, CO+H<sub>2</sub>**
- **H<sub>2</sub> Separation + Purification required**

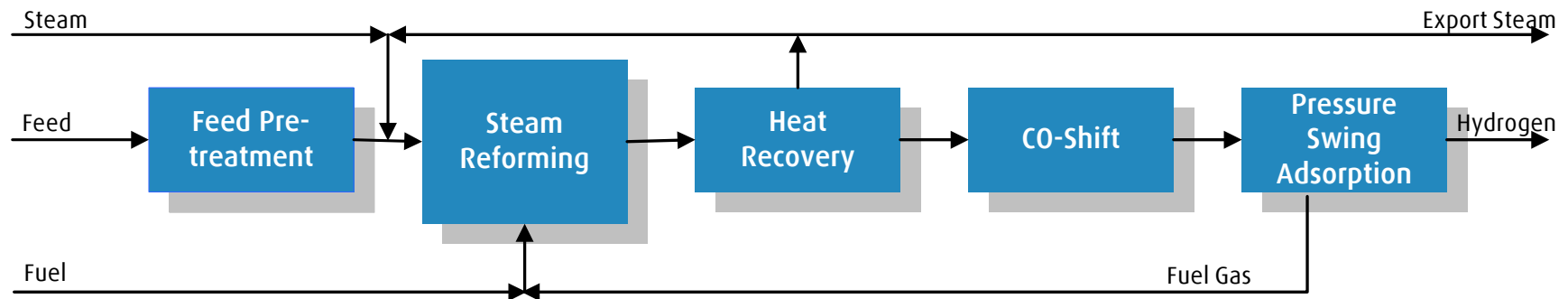
# Typical Basic Block Diagrams for H<sub>2</sub> Production



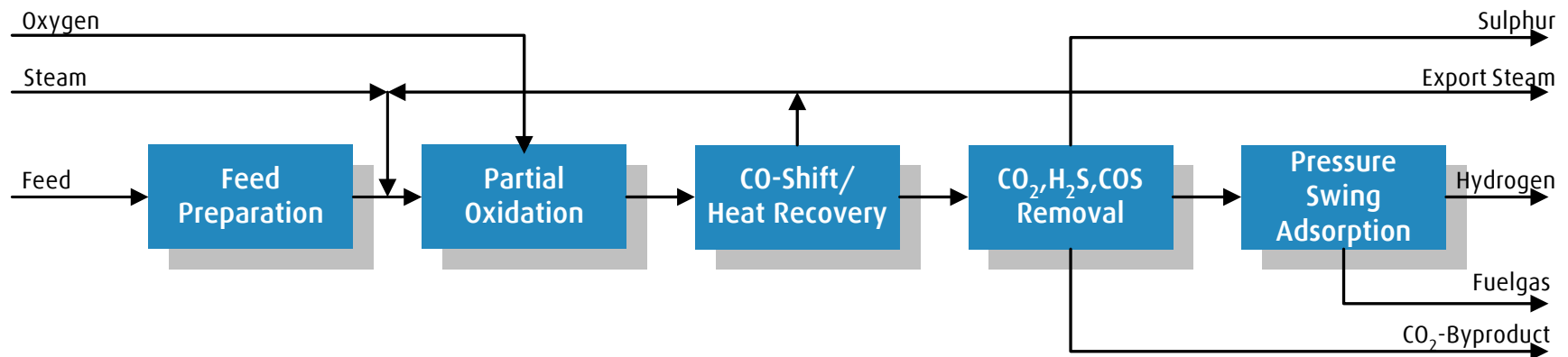
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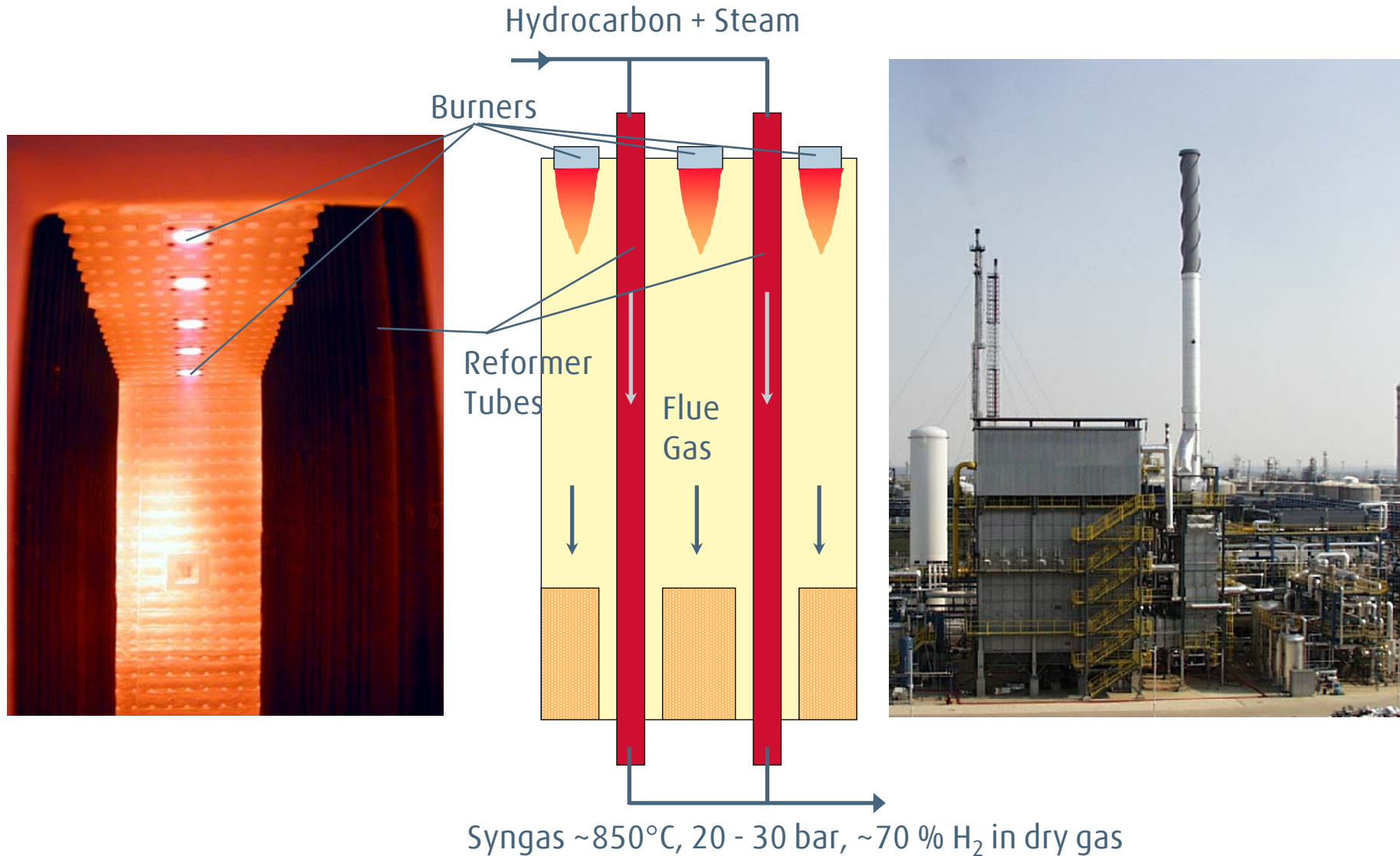
## Light Hydrocarbons



## Heavy Hydrocarbons

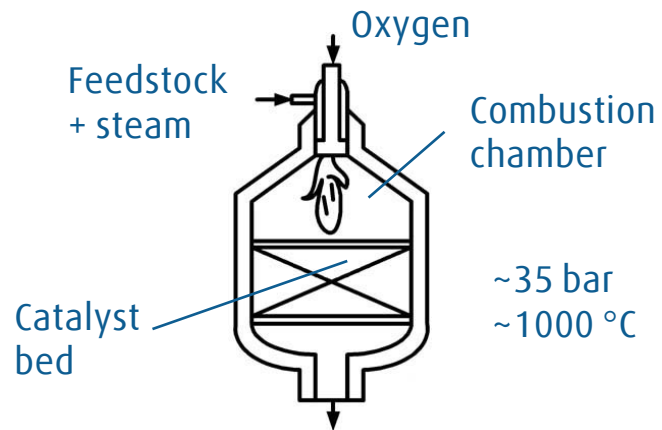


# Steam Reformer



# Partial Oxidation/ Autothermal Reforming Reactors

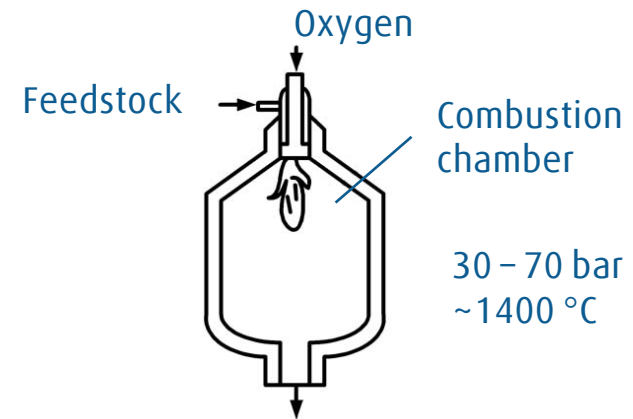
## ATR (Natural Gas)



Synthesis gas: H<sub>2</sub> in dry gas ~ 65 %



## POX (All Feedstocks)



Synthesis gas: H<sub>2</sub> in dry gas ~ 61 %



# CO-Shift Reactor

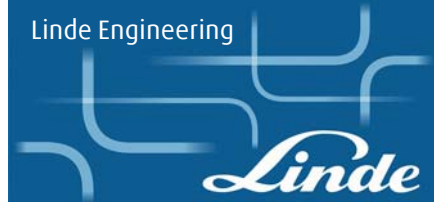
- Shifts undesired CO to H<sub>2</sub>  
 $\text{CO} + \text{H}_2\text{O} \rightarrow \text{CO}_2 + \text{H}_2$       exothermal
- Simple catalytic reactor
- CO conversion depends on Temperature  
High Temperature Shift: ~ 75 %  
Low Temperature Shift: ~ 90%
- H<sub>2</sub> in dry gas ~ 75 %



# Rectisol® Wash Unit for POX Synthesis Gas



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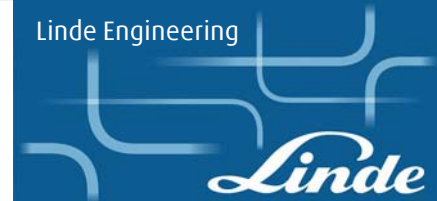
- e.g. for Syngas from Coal Gasification
- Methanol as washing solvent
- Rectisol® process separates CO<sub>2</sub>, H<sub>2</sub>S, COS
- H<sub>2</sub> Purity ~ 98 %



## H<sub>2</sub> Purification: Pressure Swing Adsorption



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- **Pressure Swing Adsorption for high Purity H<sub>2</sub>**

based on selective adsorption using different kinds of adsorption materials (e.g. molecular sieves)

- **H<sub>2</sub> Purity up to 99.9999 %**
- **H<sub>2</sub> Recovery up to 90 %**



- Major Hydrogen Market is Chemical Industry
- Feedstocks are Hydrocarbons from Methane to Coal
- Syngas Generation by Steam Reforming, Partial Oxidation, Autothermal Reforming, and CO-Shift Conversion
- H<sub>2</sub> Separation from Syngas and Purification depend on Demand and Syngas Process

Thank you  
for your attention

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